## Synthetic Cathinone Stability in Blood Using LC/Q-TOF-MS

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FSF Emerging Forensic Scientist Award Paper Presentation

### Disclosure

- There is no real or apparent conflicts of interest related to the content of this presentation
- Products used:
  - Agilent Technologies 6530 Accurate-Mass Q-TOF LC/MS
- The authors declare no competing interest
- Project funded by NIJ Award # 2013-R2-CX-K006

### **About Me**

- Originally from Chicago
- Primetime Television
  - CSI, Law and Order: SVU, Crossing Jordan, ER
- Medical Examiner
  - Medical School
- Forensic Science
- Bachelors in Chemistry from the University of Tulsa
- PhD in Forensic Science from Sam Houston State University
- Dallas County Southwestern Institute of Forensic Sciences (SWIFS)







## Synthetic Cathinone Background

- Derived from cathinone
  - Catha edulis shrub





Available on the internet and in head shops with labels such as:

"not for human consumption"

"bath salts"

"plant food"





### Pharmacology and Toxicology

#### **Desired Effects:**

- stimulant and euphoric symptoms
  - Increased energy, mood enhancement, empathy, sociability, concentration, euphoria

#### **Adverse Effects:**

- neurological, cardiovascular, and psychopathological symptoms
  - Hallucinations, delusions, confusion, violence, homicidal tendencies, death

Antemortem Cases: Motor Vehicle Accidents & Driving While Impaired

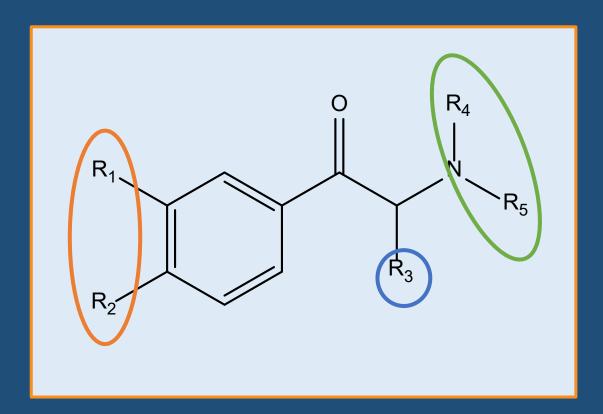
Postmortem Cases: Overdose, Suicide, Homicide





## General Structure of Cathinone

- Phenethylamines
- R groups represent positions that can be substituted to create various synthetic cathinones



### Importance of Stability

 Understanding the stability of a drug in biological matrices is essential

- Condition and length of storage can affect drug concentration
- Specimens stored for days, weeks, or months prior to analysis
- Subjected to various conditions during collection and shipping process

### Cathinone Instability

#### **Plant Material**

- Unstable in oxygen rich conditions (Szendrei, 1980)
- Unstable in alkaline conditions (Szendrei, 1980; Berrang, 1982)
- Dimer formation (Berrang, 1982; Chappell, 2010)
- Best to air dry and refrigerate

#### Thermal Degradation

- Thermal degradation in GC-MS
  - Methcathinone (DeRuiter, 1994)
  - $\blacksquare$   $\alpha$ -PVP (Tsujikawa, 2013)
  - 19 synthetic cathinones (Kerrigan, 2015)
- Aqueous Solution (Tsujikawa, 2012)
  - Stable at acidic pH
  - Decomposition rate dependent upon chemical structure
    - 5 synthetic cathinones

### **Biological Material**

#### ■ Sorensen (2011):

- Methcathinone, Ethcathinone,
   Mephedrone, Flephedrone,
   Methedrone, Methylone, Butylone
- Blood (pH 7.4 and 5.9)
- 5°C and 20°C
- 7 days
- More stable in pH 5.9 and 5°C over 7 day period

#### Johnson and Botch-Jones (2013):

- MDPV/Mephedrone
- Blood, Plasma, Urine
- -20°C, 4°C, 22°C
- 14 days
- -20°C: stable in 3 matrices
- Mephedrone unstable at 4°C and 22°C

#### Soh and Elliott (2014):

- 4-MEC
- Blood and Plasma
- 20°C and 5°C
- 7 days
- Unstable at both temperature

#### Busardo (2016):

- Mephedrone
- Antemortem & Postmortem Blood
- -20°C, 4°C, 20°C
- 6 months
- Unstable at 4°C and 20°C by 3 months
- Stable at -20°C

### Stability Studies

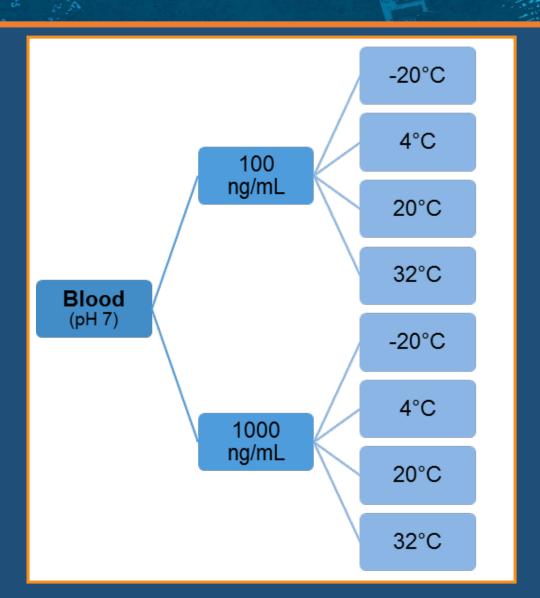
### No systematic and fully comprehensive study addressing synthetic cathinone stability in biological evidence

- 22 synthetic cathinones
- 2 biological matrix
- 4 temperatures
- >6 months

#### Comprehensive study assessing stability as it relates to

- 1. Concentration
- 2. Temperature
- 3. Storage Time
- 4. Chemical Structure

## Research Design



### LC/Q-TOF-MS Conditions

#### Agilent Technologies 6530 Accurate-Mass Q-TOF LC/MS

#### **LC Separation**

- Poroshell 120 EC-C18 Column (2.1x100mm, 2.7 μm particle size)
- Mobile Phase A: 0.1% FA in diH<sub>2</sub>O
- Mobile Phase B: 0.1% FA in ACN
- Flow Rate: 0.40 mL/min
- LC Gradient:
  - 96% A to 5 min, 90% A until 11 min, 60% A for 1 min, 0% A to equilibrate the column

#### **Q/TOF Parameters**

- Gas Temperature: 200°C
- Gas Flow Rate: 13 L/min
- Sheath Gas Temperature: 250°C
- Sheath Gas Flow Rate: 12 L/min
- Nebulizer Pressure 20 psig

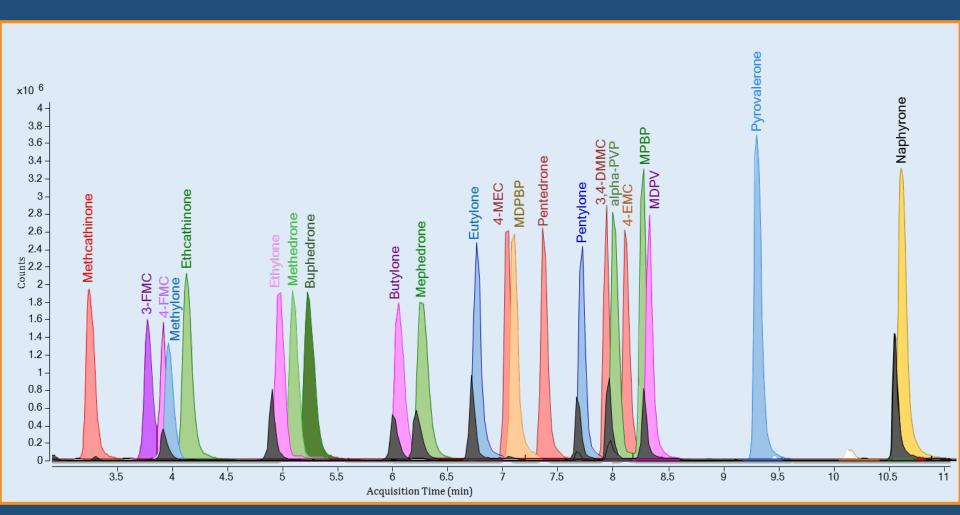
#### **Mass Spectrometry**

- Capillary Voltage: 4000 V
- Fragmentor Voltage: 150 V
- Nozzle Voltage: 0 V
- Collision Energy: 30 eV, 20 eV
- MS Scan Rate: 8 spectra/sec
- MS/MS Scan Rate: 3 spectra/sec
- MS Scan Range: 40-1000 m/z
- ESI Mode: Positive

#### **Acquisition**

- Minimum of two ion transitions per drug
- Run Time: 13 minutes

## LC/Q-TOF EIC



### Validation Summary

#### **SWGTOX Standard Practices for Method Validation**

**LOD**: 1 – 5 ng/mL

**LOQ**: 1-5 ng/mL

Inter-assay Precision: 3 - 12%

**Intra-assay Precision**: 0 – 14%

**Bias**: -7 − 11%

**Accuracy**: 93 – 100%

Matrix Effects: -15 - 3%

**Dilution Integrity**: 2- and 4- fold

Interferences: No interferences (>50 interferents)

Glicksberg, L., Bryand, K., Kerrigan, S., 2016. Identification and quantification of synthetic cathinones in blood and urine using liquid chromatography-quadrupole/time of flight (LC-Q/TOF) mass spectrometry. Journal of Chromatography B 1035, 91-103.

### Stability Study Analysis

#### **Extraction**

- Blood samples in duplicate (n=2)
  - 1000 ng/mL samples1:4 dilution
- Calibrators extracted with every run
  - 10, 25, 100, 250, 350, and 500 ng/mL
- Negative and Positive (100 ng/mL) Controls

#### **Analysis**

- Concentration Mean (n=2)
- Error bars emitted for clarity
- Significant >20% loss

Month	Samplings/Week
1	4
2-3	2-3*
4-6	1
6-12	1/month

### Secondary Amines, No Ring Substituents

Methcathinone	Ethcathinone	Buphedrone	Pentedrone
O CH <sub>3</sub>	O H N CH <sub>3</sub>	CH <sub>3</sub>	ET ZT

### Secondary Amines, Ring Substituted

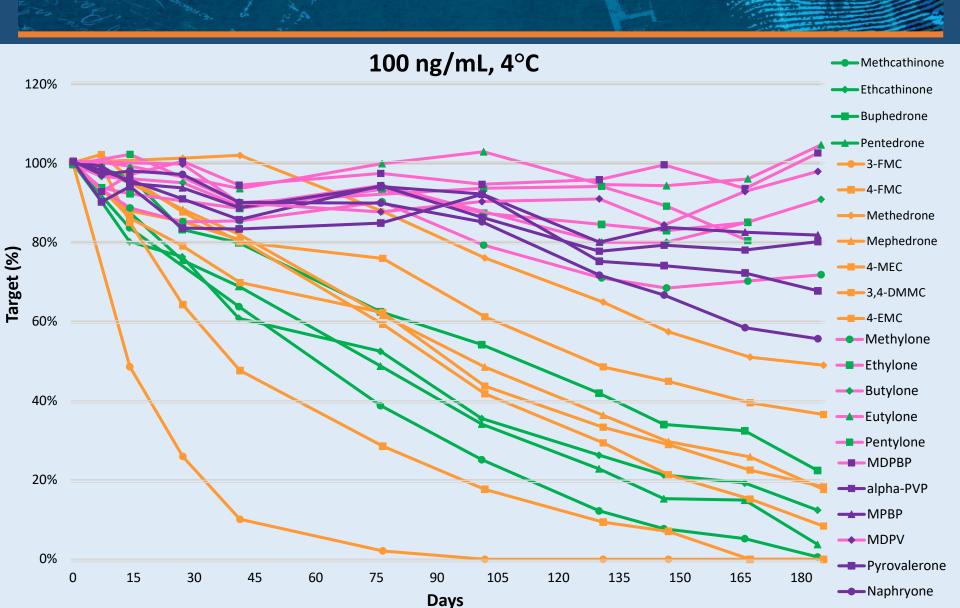
Mephedrone*	4-MEC	4-EMC	Methedrone
H <sub>3</sub> C	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C	O H CH <sub>3</sub>

### Secondary Amines, Methylenedioxy-Type

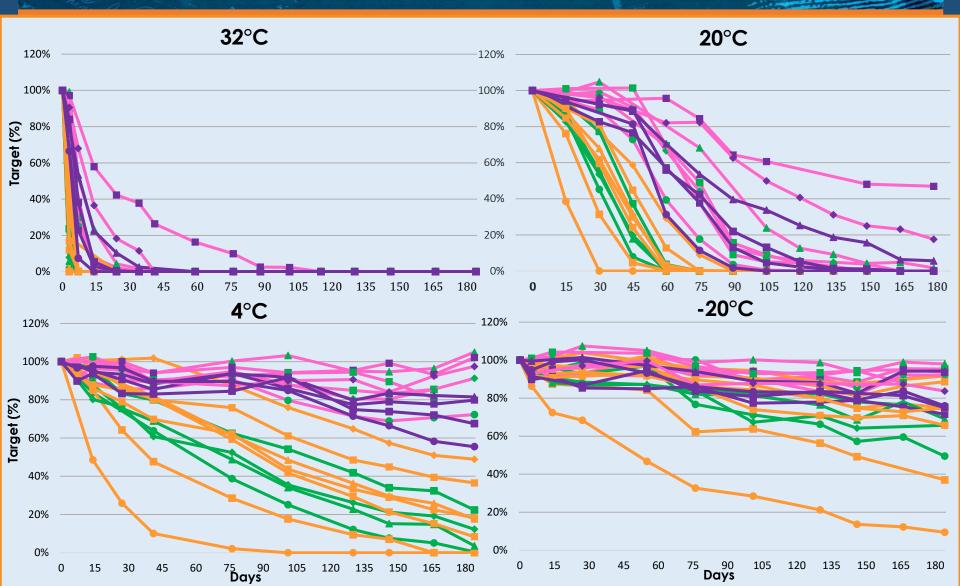
### Tertiary Amines, Pyrrolidine-Type

Alpha-PVP*	MPBP	Pyrovalerone	Naphyrone*
O CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>

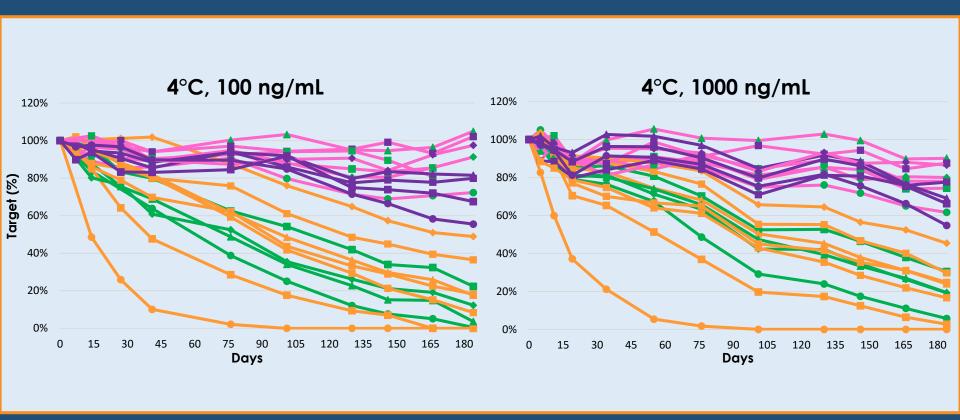
### Chemical Structure Dependence:



# Temperature Dependence (100 ng/mL)



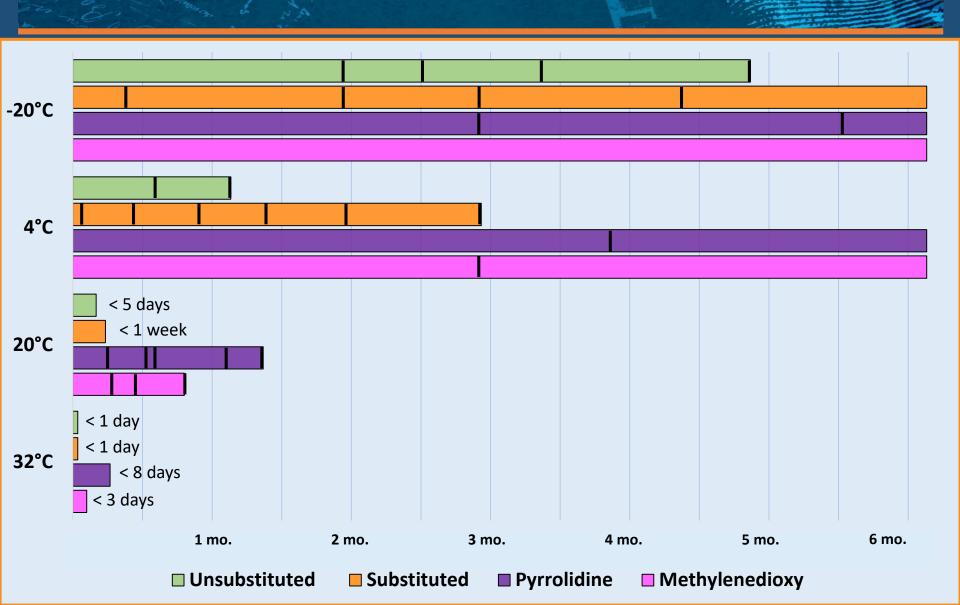
### Concentration Dependence



# Stability Range (in days)

Cathinone Structural Group	32°C	20°C	4°C	-20°C
Unsubstituted	<1	2 – 5	19 – 34	76 – 146
Ring Substituted	<1	<1-7	3 – 88	14 - >184
Methylenedioxy	1-3	9 – 24	88 - >184	>184
Pyrrolidine	2 – 8	8 – 34	115 - >184	88 - >184

### Stability Range



# Detection Window (in days)

Cathinone Structural Group	32°C	20°C	4°C	-20°C
Unsubstituted	3 – 4	19 - 24	≥184	>184
Ring Substituted	2 – 7	7 – 55	88 - >184	>184
Methylenedioxy	14 – 27	76 – 184	>184	>184
Pyrrolidine	11 – 115	55 - >184	>184	>184

### Conclusions

Temperature Dependence

-20°C > 4°C > 20°C > 32°C

No Concentration Dependence

Significant Structural Dependence

MD/PYR > PYR > MD > Ring Substituted > Unsubstituted > 3-FMC

### Conclusions Cont.

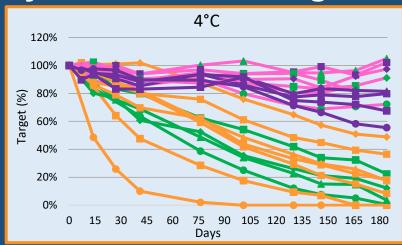
Significant loss on the order of hours

3-FMC: 32°C—undetectable after 24 hours

Substituted and Unsubstituted

32°C: >20% loss at 48 hours 20°C: >20% loss at 48 hours

Instability at common storage conditions (4°C)



Significant Structural Influence

### Acknowledgements

This project was supported by Award No. 2013-R2-CX-K006 awarded by the National Institute of Justice, Office of Justice Programs, U.S. Department of Justice. The opinions, findings, and conclusions or recommendations expressed in this publication are those of the author(s) and do not necessarily reflect those of the Department of Justice.

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### Questions?

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Friday, February 23<sup>rd</sup> @ -Post Mortem Distribution of Synthetic Cathinones

